#1: OBSERVATION

- To witness using the 5 senses (hearing, tasting, touching, smelling, sight)
- Using an instrument to extend powers of observation
 - Ex- taking measurments (data)

#2A: INFERENCE

To interpret, predict or guess based on what you observe.

#2B: Fact vs. Theory

FACT- An observable and measureable account, accepted and known to be true

THEORY-

- An explanation of how a natural process occurs that has been supported by observations over a long period of time.
- Accepted to be true, but can be changed or replaced as new data and discoveries arise.

#3: Steps of the SCIENTIFIC METHOD

2. Hypothesis(your prediction) 3. Materials (what you need) **4. Procedure**(step by step plan) **5. Observation**...(measurement, data, graph) 6. Conclusion(what you discover) 7. Error(where mistakes and outside factors effected results) "Please Help My Poor Old Cat Ernie"

#4: INDEPENDENT Variable

The "I" change variable MANIPULATED variable What was changed before the experiment starts

#5: DEPENDENT Variable

MEASURED change in experiment □What you OBSERVE Collect DATA on Responds / reacts to the **Independent Variable**

#6: Control Group vs. Experimental Group

The EXPERIMENTAL Group gets the Independent Variable. Can have multiple experimental groups

CONTROL group is exactly the same minus the independent variable

You compare the results of the experimental group to the control group.

#6B: CONSTANTS

- Variables that are "controlled"
 - Conditions kept equal
 - All factors through all test groups and parameters of the experiment must be <u>kept the SAME</u>
 - □ AKA:
 - "control factors" or
 - "controlled variables"

#6C: Graphing Variables



#7: Tools of Measurment



#8: MASS

- The amount of MATTER (atoms and molecules) in a substance.
- Measured in Grams (g)
- □ Mass is not Weight!
 - Weight is mass x gravity
 - An object has a downward force or weight due to it's mass and gravity
- Instrument -<u>TRIPLE BEAM BALANCE</u>

#9: VOLUME

The amount of SPACE an object takes up Measured in cm³ or ml Regular solid = LxWxH

Irregular solids = water displacement method

#10: DENSITY

Amount of Mass per Volume # of grams (g) in a ml or cm³ or cc $\Box D = M/V$ Units: □g/cc □g/ml □g/cm³

#10B: Density Facts

Density never changes for a pure substance

- It is a PROPERTY of matter
 - □ Ex- Aluminum is 2.7 g/cc
 - □ If you cut it in half each piece is still 2.7 g/cc
- □ Water is 1.0 g/cc
 - □ Greater then will sink
 - Less then will float

#11: Latitude and Longitude

Latitude

- Measures N and S of EQUATOR (0°)
- Lines are PARALLEL
- Goes up to 90° North and South
- Divides Earth into Northern / Southern Hemisphere

Longitude

- Measures E W of PRIME MERIDIAN (0°)
- Lines meet at the poles
- 180° (international date line) highest
- Divides Earth into Eastern /Western Hemisphere

#11b: Topographic Maps

- Each line represents an ELEVATION or height above sea-level.
- □ Lines CLOSE = STEEP gradient
- The amount each line goes up is the CONTOUR INTERVAL
- Ex:BALD HILL
 - Contour Interval 100ft

